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6 Chemical Bonding

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Chapter 6 PRETEST: Chemical Bonding

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Web liter(s), milliliter(s) g = gram(s) nm = nanometer(s) atm = atmosphere(s) mm Hg = millimeters of mercury J, kJ = joule(s), kilojoule(s) V = volt(s) mol = mole(s) ATOMIC STRUCTURE E = energy frequency wavelength Planck's constant,  $h = 6.626 \times 10^{-34}$  Js Speed of light,  $c = 2.998 \times 10^8$  ms<sup>-1</sup> Avogadro's number =  $6.022 \times 10^{23}$  mol

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